

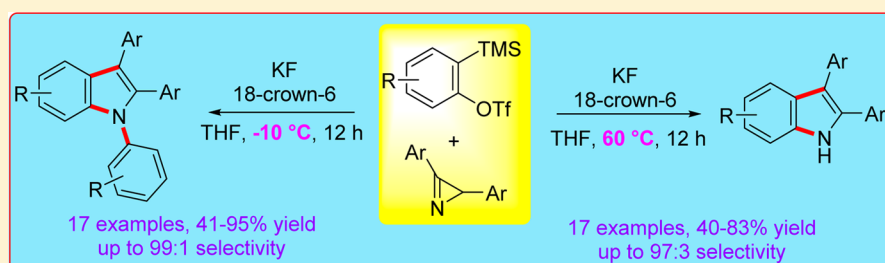
Selective Synthesis of *N*-Unsubstituted and *N*-Arylindoles by the Reaction of Arynes with Azirines

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Supporting Information



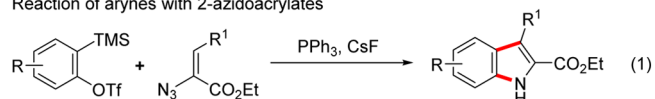
ABSTRACT: The transition-metal-free and temperature-dependent highly selective reaction of arynes with 2*H*-azirines allowing the synthesis of either *N*-unsubstituted or *N*-arylindoles has been developed. At 60 °C, arynes generated from 2-(trimethylsilyl)aryl triflates smoothly insert into 2*H*-azirines to form 2,3-diarylindoles with high selectivity. Interestingly, when the reaction was performed at −10 °C, the selectivity was switched to the formation of 1,2,3-triarylindoles in good yields.

The indole nucleus is a structural motif present in numerous biologically active natural products, pharmaceutical compounds, and agrochemicals.¹ Consequently, development of straightforward and flexible synthetic routes for the construction of functionalized indoles is of great importance.² Among the various routes to indoles, the Fischer indole synthesis stands as one of the most widely used procedures.³ Additionally, the annulation of 2-alkynylanilines under metal catalysis,⁴ cyclization of 2-halogenated anilines with alkynes,⁵ reductive annulation of 2-nitro aromatics,⁶ etc. offer convenient routes to achieve indoles. Methods employing transition-metal-catalyzed C–H bond functionalization reactions have emerged as some of the most convenient and efficient protocols for the synthesis of indoles.⁷ Furthermore, transition-metal-free transformations have also been developed recently for the synthesis of indole derivatives.⁸

The transition-metal-free synthesis of indoles employing arynes⁹ as the aryl source has been a convenient method that obviates the use of hydrazines/amines as the starting materials. The use of 2-(trimethylsilyl)aryl triflate precursor in the presence of fluoride source allows the mild method for aryne generation to proceed.¹⁰ In 2010, Wang and co-workers demonstrated the synthesis of 2,3-disubstituted indoles from arynes and 2-azidoacrylates (Scheme 1, eq 1).¹¹ In addition, Greaney and co-workers used *N*-tosylhydrazones as the coupling partner for arynes resulting in the synthesis of *N*-tosylindoles via the benzyne Fischer indole reaction (eq 2).¹² Recently, Zhu and co-workers engaged arynes in the Bischler–Möhlau synthesis of indoles, where arynes undergo an insertion–cyclization sequence

Scheme 1. Synthesis of Indoles via Arynes

Reaction of arynes with 2-azidoacrylates



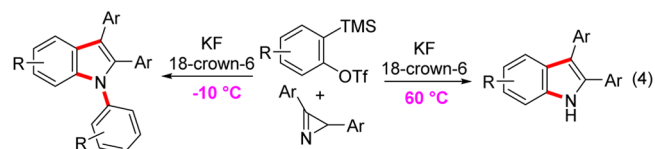
Reaction of arynes with *N*-tosyl hydrazones



Reaction of arynes with *N*-aryl α -aminoketones



Temperature dependent reaction of arynes with 2*H*-azirines (this work)



with *N*-aryl α -aminoketones (eq 3).¹³ Notably, a common method for the highly selective synthesis of both *N*-unsubstituted as well as *N*-arylindoles using aryne chemistry has not been reported.

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Herein, we report the transition-metal-free and highly selective synthesis of both *N*-unsubstituted and *N*-arylindoles by the reaction of arynes with 2*H*-azirines, and importantly, the product selectivity depends on the temperature used. When the reaction was carried out at $-10\text{ }^{\circ}\text{C}$, *N*-arylindoles were formed in high yields. Gratifyingly, when the reaction was performed at $60\text{ }^{\circ}\text{C}$, 2,3-diaryl *N*-unprotected indoles were formed in moderate to good yields and excellent selectivity. It may be noted in this context that the reaction of 2,3-diphenyl-1-azirine with arynes generated by the thermal decomposition of benzene diazonium 2-carboxylate leading to the mixture of 2,3-diphenylindole and 1,2,3-triphenylindole was reported by the Nair group as early as 1975.¹⁴ This seminal work also demonstrates the mechanism for the formation of *N*-unsubstituted and *N*-arylindoles by the reaction of arynes with azirines.

Against the literature backdrop and given the importance of functionalized indoles in organic synthesis, the present study was initiated by treating 2,3-diphenyl-1-azirine **1a** with the benzyne generated from the 2-(trimethylsilyl)aryl triflate precursor **2a** using KF in the presence of 18-crown-6 as additive in THF at $30\text{ }^{\circ}\text{C}$. Under these conditions, the 2,3-diarylindole **3a** was formed in 40% yield, and 1,2,3-triaryl indole **4a** was formed in 21% yield in a 54:46 ratio (Table 1, entry 1). Interestingly,

Table 1. Optimization of the Reaction Conditions^a

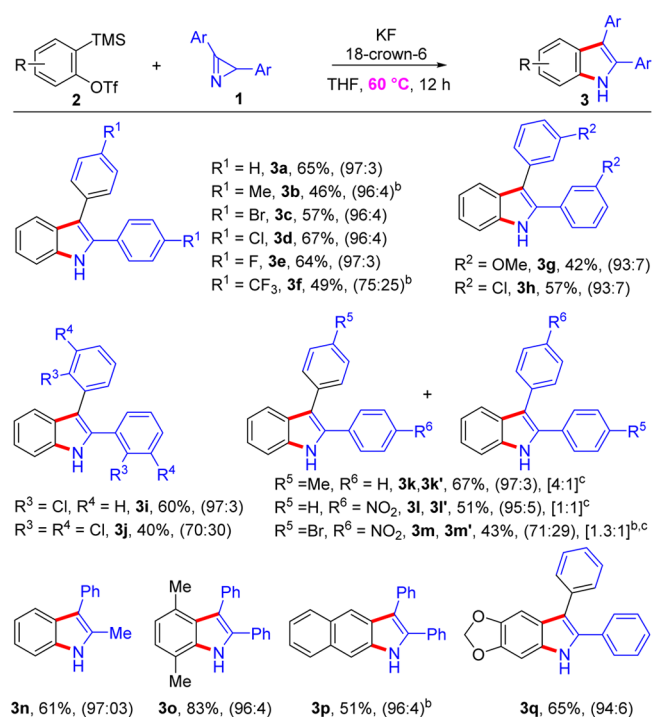
entry	fluoride source	temp ($^{\circ}\text{C}$)	yield of 3a ^b (%)	yield of 4a ^b (%)	3a : 4a ^c
1	KF/18-crown-6	30	40	21	54:46
2 ^d	KF/18-crown-6	60	65	<5	97:3
3 ^e	CsF	60	57	<5	83:17
4	TBAF.3H ₂ O	60	<5	<5	ND
5 ^f	KF/18-crown-6	-10 to $+30$	<5	76	2:98
6 ^f	KF/18-crown-6	-10	<5	83	1:99

^aGeneral conditions: **1a** (0.25 mmol), **2a** (0.30 mmol), KF (2.4 equiv), 18-crown-6 (2.4 equiv), THF (1.0 mL), for the indicated temperature and 12 h. ^bYields of the isolated products are given. ^cSelectivity was determined using GC analysis of the crude reaction mixture. ^dReaction performed using 1.8 equiv of **2a**, 3.6 equiv of KF/18-crown-6, and 6.0 mL of THF. ^eReaction performed using 1.0 mL of CH₃CN. ^fReaction carried out using 3.0 equiv of **2a**, 6.0 equiv of KF/18-crown-6, and 1.0 mL of THF.

when the reaction was performed at $60\text{ }^{\circ}\text{C}$, **3a** was isolated in 65% yield and 97:3 selectivity (entry 2). Under these conditions, only traces of **4a** were formed. The use of other fluoride sources such as CsF and tetrabutylammonium fluoride (TBAF) did not improve the yield of **3a** and **4a** (entries 3 and 4). When the reaction was carried out at $-10\text{ }^{\circ}\text{C}$ and warmed to $30\text{ }^{\circ}\text{C}$, a complete switching in selectivity from **3a** to **4a** was observed (98:2), and **4a** was isolated in 76% yield. Finally, the yield of **4a** was improved to 83% when the reaction was performed at $-10\text{ }^{\circ}\text{C}$ (entry 6).¹⁵

After optimizing the reaction conditions for the selective synthesis of *N*-unsubstituted and *N*-arylindoles, we examined the substrate scope of both transformations. First, we evaluated the scope of the synthesis of *N*-H indoles (Scheme 2). A series

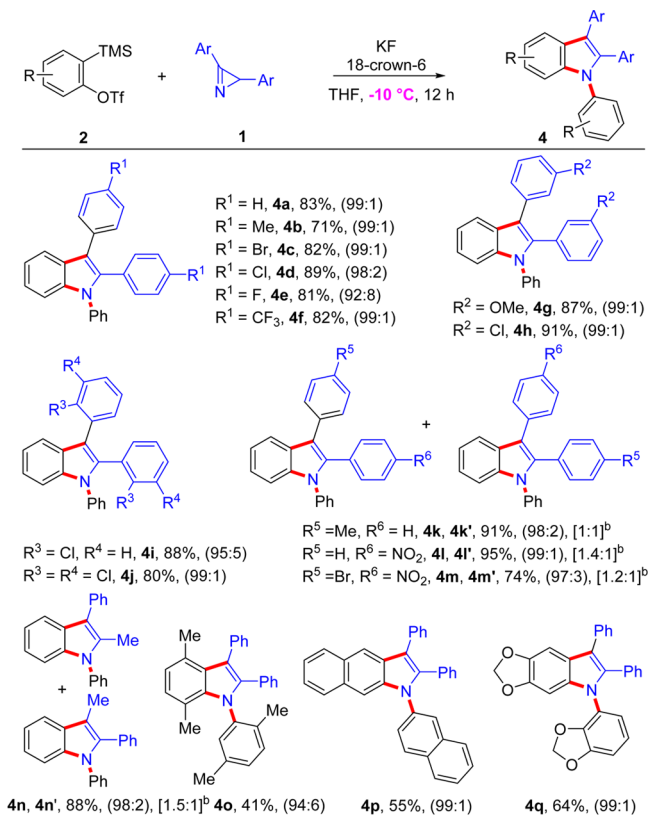
Scheme 2. Substrate Scope for the Synthesis of *N*-Unsubstituted Indoles^a



^aGeneral conditions: **1** (0.50 mmol), **2** (0.90 mmol), KF (1.80 mmol), 18-crown-6 (1.80 mmol), THF (12.0 mL), $60\text{ }^{\circ}\text{C}$, 12 h. Yields of the isolated products are given. Selectivity determined by GC analysis of crude reaction mixture is given in parentheses. ^bReaction run on 0.25 mmol scale. ^cRegioisomer ratio determined by GC analysis.

of symmetrical 2*H*-azirines bearing electron-releasing and -withdrawing groups at the 4-position of the aryl moiety of **1** are well tolerated at $60\text{ }^{\circ}\text{C}$, leading to the selective synthesis of 2,3-diarylindoles in moderate to good yield and good selectivity (**3a–f**). Notably, in the case of 4-CF₃-substituted 2*H*-azirine, the selectivity was only moderate. The structure of **3f** was further confirmed by single-crystal X-ray analysis.¹⁶ Moreover, the substitution at the 2-position and 1-position as well as disubstitution on the 2,3-diaryl moiety of **1** did not affect the arynes reactivity and furnished the desired products in moderate yield and high selectivity (**3g–j**). As expected, the unsymmetrical 2*H*-azirines afforded the inseparable mixture of regioisomers in moderate to good yield (**3k–m**). Interestingly, unsymmetrical 2-methyl-3-phenyl-2*H*-azirine afforded the single regioisomer **3n** in 61% yield and 97:3 selectivity. Additionally, symmetrical arynes generated from the corresponding precursors readily underwent smooth reaction with 2*H*-azirine **1a** to furnish the *N*-H indoles in moderate to good yields and high selectivity (**3o–q**).

With this result in hand, we then focused our attention on the synthesis of 1,2,3-trisubstituted indoles (Scheme 3). As in the case of reactions performed at $60\text{ }^{\circ}\text{C}$ for the selective synthesis of *N*-H indoles, the experiments carried out at $-10\text{ }^{\circ}\text{C}$ also showed good functional group compatibility and high selectivity toward the synthesis of trisubstituted indoles. A wide variety of symmetrical 2*H*-azirines with different substitution patterns readily underwent efficient arynes reactions leading to the synthesis of 1,2,3-triarylindoles in good yields (>71% in all cases) and excellent selectivity (>92:8 in all cases) (**4a–j**). In the case of **4h**, the structure was confirmed

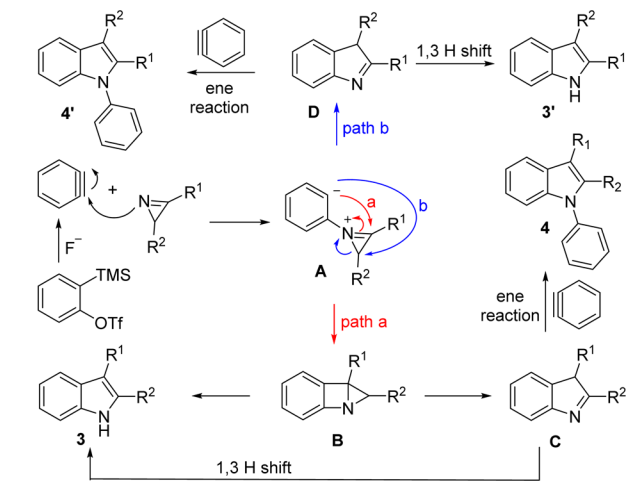
Scheme 3. Substrate Scope for Synthesis of *N*-Arylindoles^a

^aGeneral conditions: **1** (0.25 mmol), **2** (0.75 mmol), KF (1.5 mmol), 18-crown-6 (1.5 mmol), THF (1.0 mL), $-10\text{ }^\circ\text{C}$, 12 h. Yields of the isolated products are given. Selectivity as determined by GC analysis of crude reaction mixture is given in parentheses. ^bRegioisomer ratio determined by GC analysis.

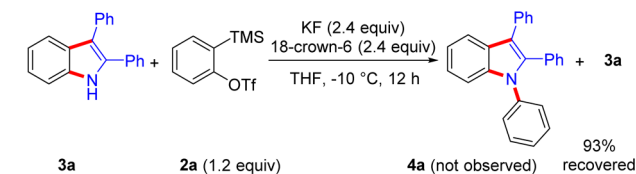
by single-crystal X-ray analysis.¹⁶ Moreover, the reaction of arynes with unsymmetrical 2*H*-azirines at $-10\text{ }^\circ\text{C}$ furnished the regioisomeric mixture of 1,2,3-triaryl indoles, where the selectivity over the *N*-*H* indoles are excellent in all cases (**4k**–**4m**). In contrast to the reaction of 2-methyl-3-phenyl-2*H*-azirine with aryne at $60\text{ }^\circ\text{C}$, the reaction at $-10\text{ }^\circ\text{C}$ afforded a regioisomeric mixture of products (1.5:1) **4n** and **4n'** in 88% yield. Furthermore, variation of the aryne moiety was also feasible, leading to the formation of the corresponding tri-substituted indoles in moderate to good yield and high selectivity, thus further expanding the scope of this aryne reaction (**4o**–**4q**).

Considering the mixture of regioisomers obtained in the reaction of arynes with unsymmetrical 2*H*-azirines, a tentative mechanism of this reaction is shown in Scheme 4. The nucleophilic addition of 2*H*-azirines onto arynes generates the 1,4 zwitterionic intermediate **A**.^{17,18} The zwitterion **A** could cyclize in two pathways. In path a, the aryl anion adds to the C=N bond (1,2-addition) to generate the intermediate **B**. A 1,2-hydrogen shift to nitrogen can result in the formation of **3**. Alternatively, a 1,2-hydrogen shift to carbon can generate the intermediate **C**, which can undergo another 1,3 hydrogen shift to afford **3**.¹⁹ An ene reaction of **C** with another molecule of aryne can furnish **4**. In path b, the aryl anion adds to carbon attached to R^2 (breaking of C–N bond) to generate the intermediate **D**. A 1,3-hydrogen shift on **D** can afford the regioisomer **3'**. Moreover, the ene reaction of **D** with another molecule of aryne can result in the formation of *N*-arylindole **4'**.

Scheme 4. Proposed Mechanism of the Reaction



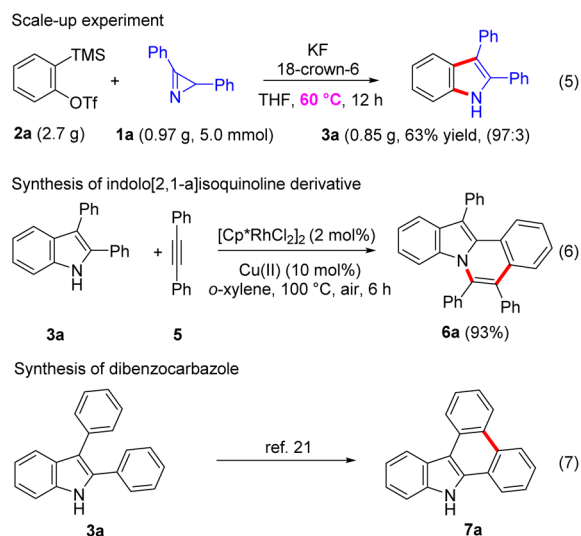
Treatment of the *N*-*H* indole **3a** with excess aryne at $-10\text{ }^\circ\text{C}$ did not afford the *N*-arylindole **4a** (Scheme 5).

Scheme 5. Attempted Reaction of **3a** with Arynes

This experiment rules out the possibility of initial formation of **3a** in the reaction of **1a** and **2a** for the synthesis of **4a**. This also sheds light on the mechanism proposed in Scheme 4.

To demonstrate the synthetic utility of the reaction, a scale-up experiment was performed. Carrying out the reaction on a 5.0 mmol scale of **1a** afforded **3a** in 63% yield and 97:3 selectivity, indicating the practical nature of the reaction (Scheme 6, eq 5).

Scheme 6. Synthetic Utility of the Method



Reaction of **3a** with alkyne **5** under Rh(III) catalysis following the procedure of Miura, Satoh, and co-workers²⁰ with a minor modification afforded the indolo[2,1-*a*]isoquinoline derivative **6a** in 93% yield (eq 6). Moreover, **3a** could easily be transformed to

5,6,12-Triphenylindolo[2,1-*a*]isoquinoline (**6a**).³³ To a 5 mL screw-capped test tube equipped with a magnetic stir bar were sequentially added 2,3-diphenyl-1H-indole (**3a**, 30 mg, 0.1 mmol), 1,2-diphenylacetylene (**5**, 24 mg, 0.13 mmol, 1.2 equiv), [(Cp**Rh*Cl₂)₂] (1.0 mg, 0.02 mmol, 2 mol %), Cu(OAc)₂·H₂O (1.0 mg, 0.1 mmol, 10 mol %), and 1 mL *o*-xylene. The tube (not sealed) was immersed in an oil bath (100 °C) and stirred vigorously under air. When TLC control showed the completion of the reaction (after 6 h), the solvent was evaporated, and subsequently, the crude residue was purified by flash column chromatography on silica gel (using pet ether–EtOAc) to afford the corresponding indolo[2,1-*a*]isoquinoline derivative **6a** as a yellow solid (0.046 g, 93% yield): *R*_f (pet ether/EtOAc = 95/05) 0.54; mp 193–195 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.01 (d, *J* = 7.9 Hz, 1H), 7.71–7.63 (m, 4H), 7.57 (d, *J* = 7.5 Hz, 2H), 7.41 (s, 5H), 7.31–7.17 (m, 9H), 6.90 (t, *J* = 7.8 Hz, 1H), 6.08 (d, *J* = 8.9 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 137.0, 136.6, 136.1, 135.7, 132.0, 131.7, 131.3, 131.2, 131.0, 130.8, 130.7, 129.2, 128.8, 128.0, 127.4, 127.1, 126.9, 126.6, 126.2, 126.0, 124.6, 121.8, 120.9, 119.1, 114.6, 112.2; HRMS (ESI) calcd [M + H]⁺ for C₃₄H₂₄N 446.1903, found 446.1900; FTIR (cm⁻¹) 3063, 1603, 1549, 1483, 1379, 1332, 1216, 1109, 1030, 924, 766, 670.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.6b01472.

- X-ray data, details on computational studies, and ¹H and ¹³C NMR spectra of all products (PDF)
- X-ray crystallographic data for **3f** (CIF)
- X-ray crystallographic data for **4j** (CIF)

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Notes

The authors declare no competing financial interest.

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(15) For a detailed computational study on the mechanism, see the Supporting Information.

(16) CCDC 1476910 (**3f**) and 1476909 (**4j**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

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